

## Cyclotrimerization of Benzobarrelene: Synthesis of New Isomeric Barrelene Architectures

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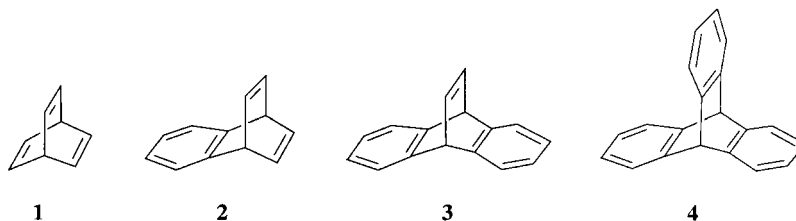
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The cyclotrimerization reaction of benzobarrelene derivatives was investigated. Dibromobenzobarrelene **10** was converted to the bromostannyl derivative **11**, which was used as the substrate of the cyclotrimerization reaction. Thus, reaction of **11**, with copper(I) thiophene-2-carboxylate (CuTC) gave a mixture of the isomeric cyclotrimers **5** and **6** and the dimers **12** and **13**, in addition to a trace of protodestannylated bromoalkene **14** (Scheme 2).

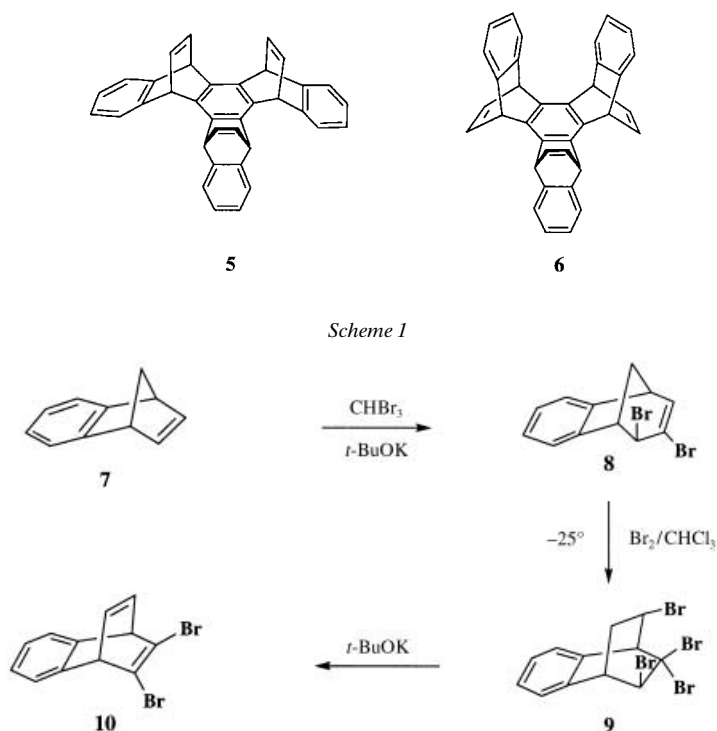
**Introduction.** – In general, the class of molecules headed by barrelene (**1**) and constituted by its benzo derivatives **2–4** offers the possibility of several mechanistically interesting investigations. For example, these compounds exhibit a unique behavior in the photochemical di- $\pi$ -methane rearrangement [1], in the cationic *Wagner–Meerwein* rearrangement [2], and in other instances [3]. In addition, they recently have raised interest as molecules related to fullerenes, and the polycyclic [2.2.2] system is often observed in fullerene derivatives [4]. From a different viewpoint, benzocyclotrimers of polycyclic systems are interesting compounds in view of the long-standing topic of aromaticity [5] and because they represent fullerene substructures or even subunits of new, hitherto unobserved carbon allotropes [6].



In this work, we present the cyclotrimerization of benzobarrelene that led to the new benzocyclotrimers **5** and **6** and gave also the opportunity to observe dimers whose symmetric and chiral structures were defined by chemical methods.

**Results and Discussion.** – The 2,3-dibromobenzobarrelene **10** was prepared by published methods [7] starting from benzonorbornadiene **7** via **8** and **9** (Scheme 1).

The reaction of dibromide **10** with BuLi followed by quenching with chlorotrimethylstannane afforded **11** in nearly quantitative yields. For the cyclotrimerization



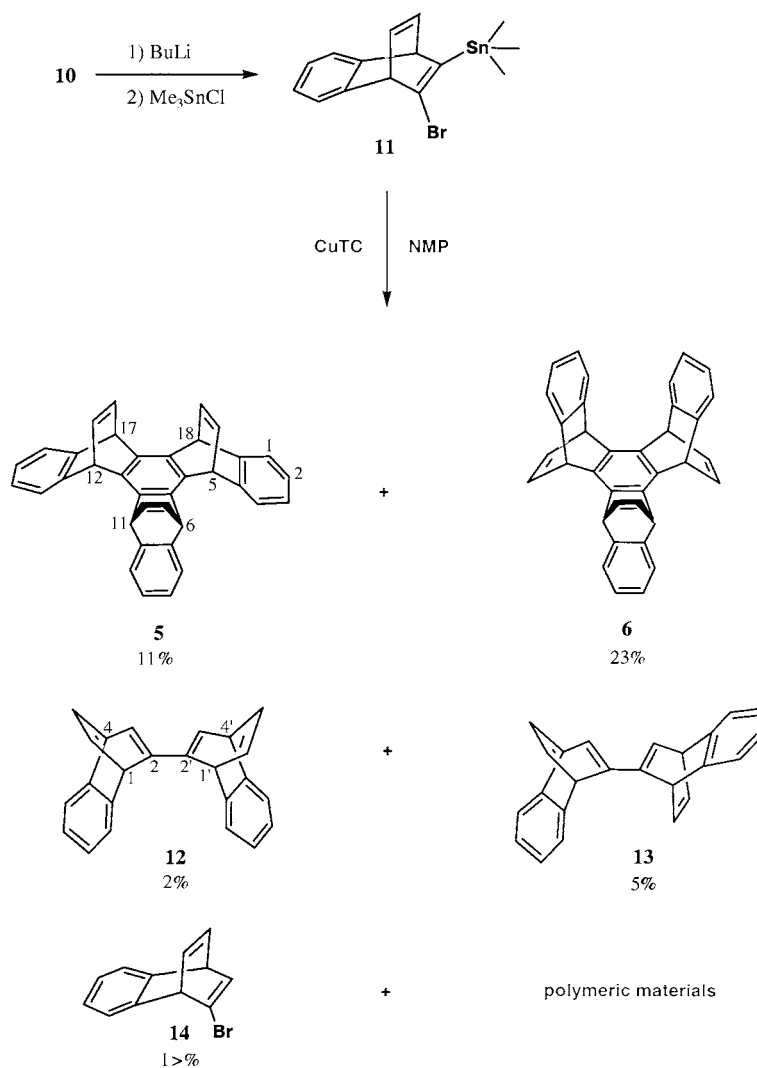
reaction, the latter was treated with copper(I) thiophene-2-carboxylate (CuTC) in 1-methylpyrrolidin-2-one as solvent at  $-20^\circ$  to provide the cyclotrimers **5** and **6** in a total yield of 34% ('*syn*'/'*anti*' ratio<sup>1)</sup> of 1:2; Scheme 2). Recently, we have demonstrated with different systems that this copper reagent provided the highest yield in the cyclotrimerization reactions [5d–g]. In the same reaction, protodebrominated dimers **12** and **13** were also detected in 7% yield in a '*syn*'/'*anti*' ratio<sup>1)</sup> of 2:5, in addition to protodestannylated product **14**.

At variance with the present reaction, where the trimers **5** and **6** were produced in a 34% total yield, the same reaction performed with vicinal bromostannyl derivatives of the norbornene system generally gave exclusively benzocyclotrimers in very high yields [5d–e]. We assumed that the reason for this observation may reside in the benzobarrelene system, *i.e.*, in both the presence of the extra C=C bond and in the diminished strain of the [2.2.2] with respect to the [2.2.1] system that cause a higher propensity to form an acetylenic intermediate, which eventually leads to polymers or undesired products. Differently, the cyclotrimerization in the [2.2.1] system proceeds *via* consecutive couplings between the functionalized C-atoms as the acetylene intermediate does not form because of its excessive strain.

The structure analyses of the trimers **5** and **6** was performed by  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectroscopy. Because the '*syn*'- and '*anti*'-isomers present different symmetry and

<sup>1)</sup> The descriptors '*syn*'/'*anti*' refer to the relative position of the etheno bridges.

Scheme 2

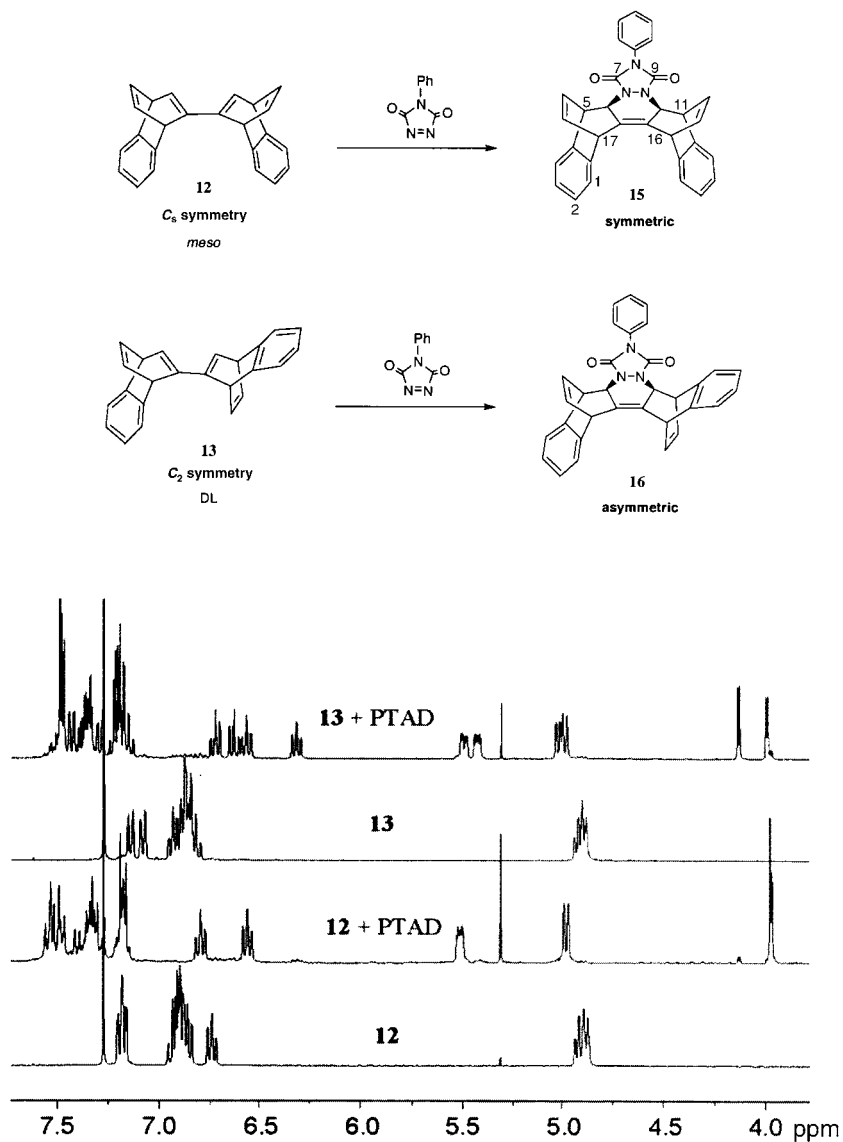


rigid structures, the NMR spectra were quite indicative, and the assignment was straightforward.

At variance, the determination of the structures of dimers **12** and **13** by spectroscopic methods was not as simple, because the  $C_s$  symmetry of the ‘*syn*’-dimer **12**, the  $C_2$  symmetry of the ‘*anti*’-isomer **13**, and the free rotation around the central  $\sigma$  bond make them indistinguishable. To determine which is which, resort was made to chemical correlation in the determination of the products of a *Diels–Alder* cycloaddition. Reaction of **12** with 1-phenyl-1*H*-1,2,4-triazole-3,5-dione (PTAD) gave a symmetrical *Diels–Alder* addition product **15**, establishing the  $C_s$  symmetry of the

starting diene and as a consequence identifying the ‘*syn*’-structure of **12**. As for the symmetric addition product **15**, it should also be noted that between the two possible *exo*- and *endo*-isomeric cycloadducts, the *exo*-addition product is exclusively formed, as is normally observed in such ‘bicyclic’ systems. At variance, the other isomer, the ‘*anti*’-dimer **13**, gave the nonsymmetric addition product **16**, as expected. These addition reactions were followed by  $^1\text{H-NMR}$  spectroscopy (Fig.).

Scheme 3

Figure. Titration experiment of **12** and **13** with PTAD in CDCl<sub>3</sub>

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### Experimental Part

*General.* All substances reported in this paper are in their racemic forms. All reactions were carried out under Ar and monitored by TLC and/or  $^1\text{H-NMR}$ . All solvents were dried and distilled before use. Flash chromatography (FC): silica gel (60 mesh, *Merck*). TLC: *Merck* 0.2-mm silica gel 60  $F_{254}$  anal. aluminium plates. M.p.: uncorrected.  $^1\text{H-}$  and  $^{13}\text{C-NMR}$  Spectra: *Varian-400* and *Bruker-300* spectrometers; apparent splitting is given in all cases.  $\delta$  in ppm,  $J$  in Hz.

(*3-Bromo-1,4-dihydro-1,4-ethenonaphthalen-2-yl*)trimethylstannane (**11**). At  $-78^\circ$ , 2.5M BuLi (2.5 ml, 6.3 mmol) was added dropwise to a soln. of dibromide **10** [7] (1.90 g, 6.1 mmol) in dry THF (20 ml), and the resulting mixture was stirred for 1 h. Chlorotrimethylstannane (1.25 g, 6.3 mmol) was added portionwise, and the mixture was stirred for 2 h at  $-78^\circ$  and then allowed to reach r.t. overnight. The crude was taken up with  $\text{H}_2\text{O}$  (50 ml) and extracted with  $\text{Et}_2\text{O}$  ( $3 \times 40$  ml), the combined  $\text{Et}_2\text{O}$  extract dried ( $\text{MgSO}_4$ ) and evaporated, and the residue purified by FC: **11** (2.36 g, 98%). Colorless liquid.  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ ): 7.20 (*m*, 1 H); 7.12 (*m*, 1 H); 6.93 (*m*, 2 H); 6.88 (*m*, 1 H); 6.81 (*dt*,  $J = 5.6, 1.5, 1$  H); 4.99 (*dd*,  $J = 5.6, 1.5, 1$  H); 4.93 (*dd*,  $J = 5.6, 1.0, 1$  H); 0.28 (*s*, 9 H).  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ ): 147.8; 145.7; 145.4; 143.9; 138.8; 138.0; 124.1; 123.6; 122.3; 121.9; 60.2; 56.1;  $-8.8$ .

*Reaction of 11 with CuTC.* To a well-stirred dispersion of copper(I) thiophene-2-carboxylate (CuTC; 1.43 g, 7.50 mmol) in dry 1-methylpyrrolidin-2-one (25 ml) at  $-20^\circ$  was added rapidly **11** (1.98 g, 5.0 mmol) *via* syringe, and the slurry was maintained at  $-20^\circ$  for 1 h. A 10% aq.  $\text{NH}_3$  soln. (20 ml) was added, and the mixture was stirred until the brown solid had disappeared. The resulting mixture was extracted with  $\text{Et}_2\text{O}$  ( $3 \times 30$  ml), the combined  $\text{Et}_2\text{O}$  extract dried ( $\text{MgSO}_4$ ) and evaporated, and the crude purified by FC (hexane/ $\text{Et}_2\text{O}$  99:1): 2-bromo-1,4-dihydro-1,4-ethenonaphthalone (**14**) [8] (10 mg, 1%), followed by 'anti'-dimer **13** (38 mg, 5%), 'syn'-dimer **12** (16 mg, 2%), 'anti'-trimer **6** (175 mg, 23%), and 'syn'-trimer **5** (85 mg, 11%). (*1RS,1'RS,4SR,4'SR*)-1,1',4,4'-Tetrahydro-2,2'-bi-1,4-ethenonaphthalene (**13**): Colorless crystals from  $\text{CH}_2\text{Cl}_2$ /hexane 1:4. M.p.  $170^\circ$ .  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ ): 7.11 (*dd*,  $J = 6.7, 1.6, 2$  arom. H); 7.09 (*dd*,  $J = 6.7, 1.6, 2$  arom. H); 6.96–6.80 (*m*, 10 H, arom. H, olef. H); 4.93 (*td*,  $J = 6.0, 1.5, 2$  aliph. H); 4.90 (*dt*,  $J = 6.0, 1.5, 2$  aliph. H).  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ ): 149.2; 147.2; 146.3; 140.2; 139.2; 130.5; 123.6; 123.4; 122.4; 121.7; 50.0; 48.6. EI-MS (70 eV): 306 (48,  $M^+$ ), 291 (30), 276 (11), 178 (100), 152 (67), 128 (56).

(*1RS,1'SR,4SR,4'RS*)-1,1',4,4'-Tetrahydro-2,2'-bi-1,4-ethenonaphthalene (**12**): Colorless crystals from  $\text{CH}_2\text{Cl}_2$ /hexane 1:3. M.p.  $225^\circ$ .  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ ): 7.19 (*m*, 4 arom. H); 6.96–6.84 (*m*, 8 H, arom. H, olef. H); 6.74 (*ddd*,  $J = 7.5, 6.0, 1.5, 2$  olef. H); 4.91 (*td*,  $J = 6.0, 1.5, 2$  aliph. H); 4.89 (*dt*,  $J = 5.7, 1.6, 2$  aliph. H).  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ ): 149.1; 147.4; 146.7; 139.8; 138.9; 130.6; 123.7; 123.5; 122.3; 121.7; 50.0; 48.5. EI-MS (70 eV): 306 (52,  $M^+$ ), 291 (30), 276 (11), 178 (100), 152 (63), 128 (51).

(*5RS,6RS,11SR,12SR,17RS,18SR*)-5,6,11,12,17,18-Hexahydro-5,18:6,11:12,17-triethenotrinaphthalene (**6**): Colorless crystals from  $\text{CH}_2\text{Cl}_2$ /hexane 1:3. M.p.  $> 310^\circ$ .  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ ): 7.31 (*AA'* of *AA'XX'*, 2 arom. H); 7.25 (*m*, 4 arom. H); 6.97 (*XX'* of *AA'XX'*, 2 arom. H); 6.93 (*m*, 8 arom. H, olef. H); 6.83 (*AA'* of *AA'XX'*, 2 olef. H); 5.50 (*m*, 6 aliph. H).  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ ): 146.2; 146.1; 146.1; 139.4; 139.2; 139.2; 134.7; 134.6; 134.5; 124.5; 124.4; 124.4; 123.1; 123.0; 122.9; 46.8; 46.7; 46.7.

(*5RS,6SR,11RS,12SR,17RS,18SR*)-5,6,11,12,17,18-Hexahydro-5,18:6,11:12,17-triethenotrinaphthalene (**5**): Colorless crystals from  $\text{CH}_2\text{Cl}_2$ /hexane 1:2. M.p.  $> 310^\circ$ .  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ ): 7.21 (*AA'* of *AA'XX'*, 6 arom. H, 7.01 (*AA'* of *AA'XX'*, 6 olef. H); 6.87 (*XX'* of *AA'XX'*, 6 arom. H); 5.50 (*XX'* of *AA'XX'*, 6 aliph. H).  $^{13}\text{C-NMR}$  (75 MHz,  $\text{CDCl}_3$ ): 146.3; 139.9; 135.0; 124.9; 123.3; 47.2.

(*5RS,5aRS,10aSR,11SR,16RS,17SR*)-5,5a,10a,11,16,17-Hexahydro-8-phenyl-5,17:11,16-dietheno-7H-benzo[*g*]naphtho[2,3-*c'*][1,2,4]triazolo[1,2-*a*]cinnoline-7(8H)-dione (**15**). To a stirred soln. of **12** (50 mg, 0.16 mmol) in  $\text{CH}_2\text{Cl}_2$  (1 ml) at  $0^\circ$ , a soln. of PTAD (30 mg, 0.17 mmol) in  $\text{CH}_2\text{Cl}_2$  (0.5 ml) was added. The red color of PTAD disappeared immediately. The solvent was evaporated, and the crude was recrystallized from  $\text{CH}_2\text{Cl}_2$ /hexane 1:1: **15** (75 mg, 96%). Colorless crystals. M.p.  $208-210^\circ$  (dec.).  $^1\text{H-NMR}$  (300 MHz,  $\text{CDCl}_3$ ): 7.54–7.17 (*m*, 13 arom. H); 6.80 (*ddd*,  $J = 7.7, 6.3, 1.6, 2$  olef. H); 6.57 (br. *dd*,  $J = 7.5, 6.3, 2$  olef. H); 5.51 (*ddd*,  $J = 6.3, 3.2, 1.6, 2$  aliph. H); 4.99 (br. *d*,  $J = 6.3, 2$  aliph. H); 3.98 (br. *d*,  $J = 3.2, 2$  aliph. H).  $^{13}\text{C-NMR}$

(75 MHz, CDCl<sub>3</sub>): 150.0; 140.4; 140.2; 135.4; 132.5; 131.4; 129.0; 128.8; 128.1; 126.7; 126.5; 125.9; 124.8; 123.6; 57.6; 43.6; 43.5.

(5RS,5aRS,10aSR,11RS,16SR,17SR)-5,5a,10a,11,16,17-Hexahydro-8-phenyl-5,17:11,16-dietheno-7H-benzo[*g*]naphtho[2,3-*c*][1,2,4]triazolo[1,2-*c*]cinnoline-7,9(8H)-dione (**16**). As described for **15**, with **13** (50 mg, 0.16 mmol) and PTAD (30 mg, 0.17 mmol): **16** (75 mg, 96%). Colorless crystals. M.p. 175–177° (dec.). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>): 7.50–7.13 (*m*, 13 arom. H); 6.73 (*ddd*, *J* = 7.4, 6.0, 1.3, 1 olef. H); 6.64 (*ddd*, *J* = 7.4, 6.5, 1.4, 1 olef. H); 6.58 (*ddd*, *J* = 7.4, 6.4, 1.5, 1 olef. H); 6.33 (*br. dd*, *J* = 7.4, 6.3, 1 olef. H); 5.50 (*ddd*, *J* = 6.5, 2.7, 1.4, 1 aliph. H); 5.43 (*ddd*, *J* = 6.5, 2.7, 1.5, 1 aliph. H); 5.03 (*dd*, *J* = 6.0, 1.3, 1 aliph. H); 4.99 (*br. d*, *J* = 6.4, 1 aliph. H); 4.14 (*d*, *J* = 2.7, 1 aliph. H); 4.00 (*d*, *J* = 2.7, 1 aliph. H). <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>): 150.1; 150.1; 142.3; 140.3; 140.3; 138.3; 134.8 (2C); 134.2; 132.3; 131.4; 129.2; 129.0; 128.0; 127.9; 126.63; 126.59; 126.5; 126.3; 126.1; 125.8; 124.8; 123.6; 121.5; 57.5; 56.9; 43.4; 43.4 (2 C); 42.9.

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